

Preliminary communication

A total synthesis of glycononaosyl ceramide with a sialyl dimeric Le^x sequence [☆]

Masami Iida ^a, Akira Endo ^a, Shuji Fujita ^a, Masaaki Numata ^a,
Yuji Matsuzaki ^a, Mamoru Sugimoto ^a, Shigeki Nunomura ^{a,*},
Tomoya Ogawa ^{b,c}

^a Tokyo Research Institute, NISSIN Food Products Co., Ltd., 1780 Kitano, Tokorozawa-shi,
Saitama 359, Japan

^b The Institute of Physical and Chemical Research (RIKEN) Wako-shi, Saitama 351-01, Japan

^c Department of Cellular Biochemistry, University of Tokyo, Yayoi, Bunkyo-ku, Tokyo 113, Japan

Received 30 November 1994; accepted 17 February 1995

Keywords: Ceramide, glycononaosyl; Lewis X (Le^x), sialyl dimeric.

E-selectin (endothelial-leukocyte adhesion molecule-1), P-selectin (GMP-140) and L-selectin (leukocyte adhesion molecule-1) belong to a family of adhesion molecules that mediate the binding of leucocytes to endothelial cells and platelets, as well as to lymphocyte-homing receptors. The ligand recognized by E-selectin is the SLe^x type determinant [1] which is found as the terminal carbohydrate structure in both glycolipids and glycoproteins. From this background, Hasegawa and co-workers have reported the synthesis of sialyl Le^x gangliosides and analogues to clarify structure–activity relationships in this epitope [2].

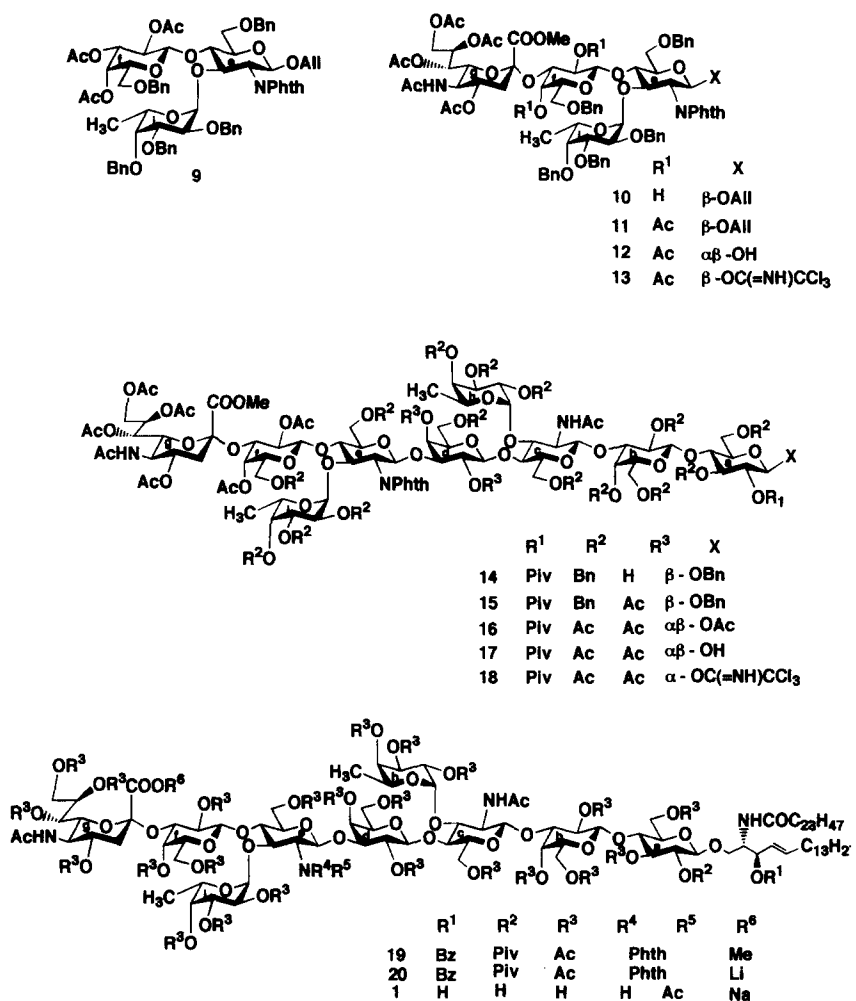
Sialyl dimeric Le^x exhibits high potency among the naturally occurring E-selectin binding molecules, but nobody has reported its synthesis as the naturally occurring glycolipid structure due to its molecular complexity, although the carbohydrate portion was synthesized by Nicolaou et al. [3]. It is noteworthy that glycononaosyl ceramide **1** is identified as tumor-associated ganglioside that accumulates in human colonic adenocarcinoma but is absent in normal colonic mucosa [4]. Owing to the biological importance of **1**, an efficient chemical synthesis is in demand.

[☆] For a preliminary report, see M. Iida, S. Nunomura, M. Numata, M. Sugimoto, K. Tomita, and T. Ogawa, *Nippon Nōgeikagaku Kaishi* (Japan Society for Bioscience, Biotechnology, and Agrochemistry), 67 (1993) 181.

* Corresponding author.



Glycosylation of **6** [8] (3 equiv) with **7** in CH₃CN under the influence of HgBr₂-Hg(CN)₂ at 0°C afforded a 41% yield of the desired α-(2 → 3)-linked tetrasaccharide **10**



Scheme 2.

$[\alpha]_D - 4.9^\circ$ (c 0.3); R_f 0.14 (10:1 toluene–MeOH); mp. 193°C]¹ along with 19% yield of its β -isomer. Compound **10** was easily separated from its β -isomer by recrystallization from MeOH. The regiochemistry of the newly introduced glycosidic linkage of **10** was deduced by converting **10** into acetate **11** [$[\alpha]_D - 9.1^\circ$ (c 0.2); R_f 0.25 (9:1 toluene–MeOH)], which showed in the homonuclear Hartmann–Hahn

¹ It should be noted that all new compounds described herein gave satisfactory elemental analyses. Optical rotations were determined for solutions in CHCl₃ at 22°C . NMR spectra were recorded with a JNM-GX 500 Fourier-transform instrument.

The values of δ_H are expressed in ppm downfield from the signal for internal Me₄Si for solutions in CDCl₃ at 25°C , unless noted otherwise. Mass spectra were determined using electrospray-ionization (ESIMS) and fast-atom bombardment mass spectroscopy (FABMS) techniques.

(HOHAHA) NMR spectrum newly deshielded signals for H-2f at δ_{H} 4.754 (dd, $J = 8.8, 10.2$ Hz), H-4f at δ_{H} 5.058 (d, $J = 3.3$ Hz). The observed chemical shifts and coupling constants of the Neu5Ac unit for H-3g (δ_{H} 2.542, $J_{3\text{a},3\text{e}} = 12.5$ Hz, $J_{3\text{e},4} = 4.8$ Hz), H-4g (δ_{H} 4.908), and H-7g (δ_{H} 5.391, $J_{6,7} = 2.6$ Hz, $J_{7,8} = 9.5$ Hz) are characteristic of the α -glycosidic linkage of Neu5Ac, in good agreement with previous observations [9].

Deallylation of **11** with (1) $[\text{Ir}(\text{COD})(\text{PMePh}_2)_2]\text{PF}_6$ [10] in THF and (2) I_2 in aq THF gave hemiacetal **12** in 84% yield. Compound **12** was transformed into β -trichloroacetimidate **13** (R_f 0.33 (3:1 CHCl_3 –acetone); δ_{H} 6.394 (d, $J = 8.8$ Hz, H-1e)) in 80% yield in the presence of CCl_3CN and 1,8-diazabicyclo [5,4,0]undec-7-ene (DBU) [11]. Having prepared the designed tetrasaccharide donor **13** and the glycosyl acceptor **8** [6], crucial glycosylation was examined. Boron trifluoride etherate-promoted glycosylation of **13** (1.5 equiv) with **8** in CH_3CN at -40°C was successfully achieved in a regio- and stereo-controlled manner to give **14** in 52% yield: ($[\alpha]_{\text{D}} -23^\circ$ (c 1.0); R_f 0.48 (3:1 CHCl_3 –acetone)). But in the case of trimethylsilyl triflate-promoted glycosylation [12], the coupling yield was decreased to 30%. The configuration of the newly introduced anomeric carbon C-1e was expected to be β , due to the presence of the *N*-2 phthaloyl group in the glycosyl donor that favors the formation of 1,2-*trans* stereochemistry. Indeed the ^1H NMR spectral data showed the anomeric proton of H-1e at δ_{H} 5.234 (d, $J = 8.4$ Hz), thus confirming the β configuration. The regiochemistry of **14** was deduced by converting **14** into acetate **15** ($[\alpha]_{\text{D}} -19^\circ$ (c 0.8); R_f 0.7 in (3:1 CHCl_3 –acetone))₂, which showed in the ^1H NMR spectrum a newly deshielded signal for H-4d at δ_{H} 5.470 (d, $J = 4.0$ Hz) and H-2d at δ_{H} 4.616 (dd, $J = 8.1, 9.2$ Hz).

Conversion of **15** into the completely acylated glycononaose **16** was carried out in two steps in 99% overall yield as follows: (1) H_2 with 20% $\text{Pd}(\text{OH})_2\text{-C}$ in 4:1 $\text{MeOH-H}_2\text{O}$; (2) Ac_2O and 4-(dimethylamino)pyridine (DMAP) in pyridine. Compound **16** was obtained as a 1:1 mixture of α : β anomers at C-1a [R_f 0.36 (20:1 CHCl_3 – MeOH); δ_{H} 6.284 (d, $J = 3.7$ Hz, H-1a α) and δ_{H} 5.688 (d, $J = 8.1$ Hz, H-1a β)]. Chemoselective cleavage of the anomeric acetate of **16** with hydrazinium acetate [13] in DMF at room temperature gave **17** in 98% yield. Compound **17** was treated with CCl_3CN and DBU in $(\text{ClCH}_2)_2$ to afford α -trichloroacetimidate **18** in 91% yield: (R_f 0.33 (25:1 CHCl_3 – MeOH); δ_{H} 6.489 (d, $J = 3.7$ Hz, H-1a)). The crucial coupling between **18** and **3** was performed in freshly distilled CHCl_3 in the presence of boron trifluoride etherate at -15°C to afford a 39% yield of β -glycoside **19** ($[\alpha]_{\text{D}} -30.6^\circ$ (c 1.0); R_f 0.38 (1:1 toluene–acetone)). The newly formed glycosidic linkage was shown to be β as revealed in the HOHAHA NMR spectrum of **19** (δ_{H} 4.403 (d, $J = 7.7$ Hz, H-1a)). Further conversion of **19** to the target glycolipid **1** was executed as follows. Compound **19** was refluxed for 7 h with a large excess of LiI in dry pyridine [14]. Purification of the reaction mixture by gel filtration through Sephadex LH-20 in 1:2 CHCl_3 – MeOH gave a 93% yield of the lithium salt **20**. Subsequent treatment of **20** with (1) NH_2NHMe in refluxing EtOH , (2) Ac_2O in MeOH and (3) aq NaOH in 1:1 MeOH-THF , afforded the target compound **1** (4.0 mg) in 47% yield, after gel filtration through Sephadex LH-20 using 5:5:1 CHCl_3 – $\text{MeOH-H}_2\text{O}$.

Physicochemical data for **1**: R_f 0.14 (2:1:1 $\text{BuOH-EtOH-H}_2\text{O}$); ^1H NMR (50:1 $\text{Me}_2\text{SO-}d_6\text{-D}_2\text{O}$, 60°C): δ_{H} 5.601 (dt, 1 H, $J = 15.4, 7.0$ Hz, H-5Cer), 5.410 (dd, 1 H,

$J = 15.4$, 7.0 Hz, H-4Cer), 4.935 (d, 2 H, $J = 3.3$ Hz, H-1h and H-1i), 4.802 (d, 1 H, $J = 7.3$ Hz, H-1c or H-1e), 4.788 (d, 1 H, $J = 6.6$ Hz, H-1e or H-1c), 4.404 (d, 1 H, $J = 7.3$ Hz, H-1d or H-1f), 4.360 (d, 1 H, $J = 7.7$ Hz, H-1f or H-1d), 4.334 (d, 1 H, $J = 7.0$ Hz, H-1b), 4.220 (d, 1 H, $J = 8.1$ Hz, H-1a), 2.818 (dd, 1 H, $J = 5.1$, 12.1 Hz, H-3g_{eq}), 1.169 (d, 3 H, $J = 6.6$ Hz, H-6h or H-6i), 1.086 (d, 3 H, $J = 6.6$ Hz, H-6i or H-6h), 1.945, 1.881, and 1.875 (3 s, 9 H, 3 NAc), 0.897 (t, 6 H, $J = 7.0$ Hz, 2CH₂Me); ESIMS: m/z (M + 2Na)²⁺ 1177.9; FABMS (TEA matrix): m/z (M – Na)[–] 2287.

The biological properties of **1** are currently being studied. In summary, an unambiguous total synthesis of sialyl dimeric Le^x glycononaosyl ceramide was achieved for the first time in a regio- and stereo-controlled manner using **18** as a key glycosyl donor.

Acknowledgments

We thank Mr Tadashi Ii and Dr Yoko Ohashi of the Institute of Physical and Chemical Research for recording the ESI and FAB mass spectra.

References

- [1] J.B. Lowe, L.M. Stoolman, R.P. Nair, R.D. Larsen, T.L. Berhend, and R.M. Marks, *Cell*, 63 (1990) 475–484; M.L. Phillips, E. Nudelman, F.C.A. Gaeta, M. Perez, A.K. Singhal, S. Hakomori, and J.C. Paulson, *Science*, 250 (1990) 1130–1132; G. Walz, A. Aruffo, W. Kolanus, M. Bevilacqua, and B. Seed, *Science*, 250 (1990) 1132–1135.
- [2] A. Kameyama, H. Ishida, M. Kiso, and A. Hasegawa, *Carbohydr. Res.*, 209 (1991) C1–C4; *J. Carbohydr. Chem.*, 10 (1991) 549–560; *J. Carbohydr. Chem.*, 10 (1991) 729–738; A. Hasegawa, T. Ando, A. Kameyama, and M. Kiso, *Carbohydr. Res.*, 230 (1992) C1–C5; *J. Carbohydr. Chem.*, 11 (1992) 645–658; A. Hasegawa, T. Ando, M. Kato, H. Ishida, and M. Kiso, *Carbohydr. Res.*, 257 (1994) 67–80; for a synthesis of a closely related compound, see T. Ehara, A. Kameyama, Y. Yamada, H. Ishida, M. Kiso, and A. Hasegawa, *Abstr. Papers 17th Int. Carbohydr. Symp.*, 17 (1994) 250.
- [3] K.C. Nicolaou, C.W. Hummel, and Y. Iwabuchi, *J. Am. Chem. Soc.*, 114 (1992) 3126–3128.
- [4] H. Nakasaki, T. Mitomi, T. Noto, K. Ogoshi, H. Hanaue, Y. Tanaka, H. Makuuchi, H. Clausen, and S. Hakomori, *Cancer Res.*, 49 (1989) 3662–3669.
- [5] K. Koike, Y. Nakahara, and T. Ogawa, *Glycoconj. J.*, 1 (1984) 107–109; K. Koike, M. Numata, M. Sugimoto, Y. Nakahara, and T. Ogawa, *Carbohydr. Res.*, 158 (1986) 113–123.
- [6] S. Nunomura, M. Iida, M. Numata, M. Sugimoto, and T. Ogawa, *Carbohydr. Res.*, 263 (1994) C1–C6.
- [7] S. Sato, S. Nunomura, T. Nakano, Y. Ito, and T. Ogawa, *Tetrahedron Lett.*, 29 (1988) 4097–4100; S. Sato, Y. Ito, and T. Ogawa, *Tetrahedron Lett.*, 29 (1988) 5267–5270; S. Nunomura and T. Ogawa, *Tetrahedron Lett.*, 29 (1988) 5681–5684.
- [8] R. Kuhn, P. Lutz, and D.L. MacDonald, *Chem. Ber.*, 99 (1966) 611–617.
- [9] H. Paulsen and H. Tietz, *Angew. Chem., Int. Ed. Engl.*, 21 (1982) 927–928; *Carbohydr. Res.*, 125 (1984) 47–64; K. Okamoto, T. Kondo, and T. Goto, *Tetrahedron Lett.*, 27 (1986) 5229–5236.
- [10] L.M. Haines and E. Singleton, *J. Chem. Soc., Dalton Trans.*, (1972) 1891–1896; J.J. Oltvoort, C.A.A. van Boeckel, J.H. De Koning, and J.H. van Boom, *Synthesis*, (1981) 305–308.
- [11] R.R. Schmidt and J. Michel, *Angew. Chem., Int. Ed. Engl.*, 19 (1980) 731–732.
- [12] H. Vorbrüggen and K. Krolikiewicz, *Angew. Chem., Int. Ed. Engl.*, 14 (1975) 421–422; S. Murata, M. Suzuki, and R. Noyori, *Tetrahedron Lett.*, 21 (1980) 2527–2528; T. Ogawa, K. Beppu, and S. Nakabayashi, *Carbohydr. Res.*, 93 (1981) C6–C9.
- [13] G. Excoffier, D. Gagnaire, and J.-P. Utile, *Carbohydr. Res.*, 39 (1975) 368–373.
- [14] E. Taschner and B. Liberek, *Rocz. Chem.*, 30 (1956) 323–325; *Chem. Abstr.*, 51 (1957) 1039d; F. Elsinger, J. Schreiber, and A. Eschenmoser, *Helv. Chim. Acta*, 43 (1960) 113–118; M. Sugimoto, M. Numata, K. Koike, Y. Nakahara, and T. Ogawa, *Carbohydr. Res.*, 156 (1988) C1–C5.